GOVERNMENT COLLEGE, ISRANA LESSON PLAN

Department of Chemistry

Session 2024-25, odd semester

Name of Assistant Professor: Sh. KULBIR KADYAN

B.Sc. 1st year	1 Disab Flamenta Salient features of	Group discussion
July-August	s and p-Block Elements Salient features of hydrides, oxides, halides, hydroxides of sblock elements (methods of preparation excluded). Structure, preparation and properties of Diborane and Borazine. Catenation, carbides, fluorocarbons, silicates (structural aspects), structure of oxides of Nitrogen and Phosphorous, structure of white and red phosphorous. Structure of oxyacids of Nitrogen, phosphorous, sulphur and chlorine and comparison of acidic strength of oxyacids. low chemical reactivity of noble gases, chemistry of xenon, structure and bonding in fluorides, oxides and oxyfluorides of xenon.	
September	Electrochemistry-I Electrolytic conduction, factors affecting electrolytic conduction, specific conductance, molar conductance, equivalent conductance and relation among them, their variation with concentration. Application of Kohlrausch's Law in calculation of conductance of weak electrolytes at infinite dilution (Numericals) Concepts of pH and pKa, Buffer solution, Buffer action, Henderson – Hazel equation, Buffer mechanism of buffer action. Reversible & irreversible cells, Calculation of thermodynamic quantities of cell reaction (AG, AH & K). Types of reversible electrodes – metal- metal ion, gas electrode, metal – insoluble salt- anion and redox electrodes. Nernst equation, Standard Hydrogen electrode, reference electrodes, Applications of EMF measurement in solubility product and potentiometric titrations using glass electrode.	Group discussion

October	Alkynes Nomenclature and its structure.	Group discussion
	Methods of formation: using Calcium	, a
	carbide, dehydrohalogenation, Kolbe's	
	electrolysis. Chemical reactions:	
	Mechanism of electrophilic and	
	nucleophilic addition reactions, formation	
	of metal acetylides, addition of bromine	
	and alkaline KMnO4, ozonolysis. Acidity	
	of alkynes. Stereochemistry of Organic	
	Compounds Concept of isomerism:	
	Structural and Stereoisomerism. Symmetry	v.
	elements, enantiomers, optical activity,	⊎
	properties of chambers,	
	achiral molecules (up-to 2 asymmetric	
	centres), diastereomers, threo- and	
	erythronomenclature, meso-compounds,	
	Relative and absolute configuration,	
	sequence rules, R and S system of	
	nomenclature. Cis- Trans isomerism, E &	
	Z system of nomenclature, Conformational	
	analysis of ethane and n-butane,	
	conformations of cyclohexane, axial and	,
	equatorial bonds. Newman and Sawhorse	
	projection formulae.	
November	Benzene and its derivatives:	Group discussion
	Nomenclature, Aromatic nucleus and side	-
	chain, Huckels' rule of aromaticity.	
	chain, Huckels' rule of aromaticity. Aromatic electrophilic substitution.	
	Aromatic electrophilic substitution,	
	Aromatic electrophilic substitution, mechanism of nitration, halogenation,	
	Aromatic electrophilic substitution, mechanism of nitration, halogenation, sulphonation, and Friedel- Crafts reaction.	
	Aromatic electrophilic substitution, mechanism of nitration, halogenation, sulphonation, and Friedel- Crafts reaction. Energy profile diagrams. Activating,	
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	polyethene pollution, pollution sources,	
	Recycling of plastic, greenhouse effect,	
	Group discussion ozone depletion	
September	Energy: Energy sources, renewable and	Group discussion
september	non-renewable sources, cells and batteries,	
0 ()	fuel cell, solar cell, polymer cell	Group discussion
October	Water: Sources of drinking water and uses,	Group discussion
	water conservation, Permissible TDS,	
	Techniques of purification of water, R.O.	
	water purification process (Osmosis and	
	Reverse Osmosis), wastewater	
	management	
November	Pesticides and Herbicides: General	Group discussion
	introduction and definition, biological	
	control and chemical control: natural and	
	synthetic pesticides, benefits and adverse	
	effects of DDT, BHC, malathion.	
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	Canhahadaataa	Group discussion
July-August	Carbohydrates Classification and nomenclature of	Group discussion
	Monosaccharides, mechanism of osazone	
	formation, interconversion of glucose and	
	fructose, chain lengthening and chain	
	shortening of aldoses. Configuration of	
	monosaccharides. Erythro and	
	threodiastereomers. Conversion of glucose	
	into mannose. Formation of glycosides,	
	Determination of ring size of glucose and	
	fructose. Open chain and cyclic structure	
	of D(+)-glucose & D(-) fructose.	
	Mechanism of mutarotation.	0
September-	NMR Spectroscopy	Carana di anni
October		Group discussion
October	Principle of nuclear magnetic resonance,	
	the PMR spectrum, number of signals,	t and the second
	peak areas, equivalent andnonequivalent	
	protons positions of signals and chemical	
	shift, shielding and deshielding of protons.	
	proton counting, splitting of signals and	
	coupling constants, magnetic equivalence	
	of protons. Discussion of PMR spectra of	
	themolecules: ethyl beautif	
	themolecules: ethyl bromide, n-propyl	
	bromide, isopropyl bromide, 1,1-	
	dibromoethane, ethanol, acetaldehyde	
	ethyl acetate, toluene, benzaldehyde and	}
	acetophenone. Simple problems on	
	PMRspectroscopy for structure	

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November	Organometallic Compounds Organomagnesium compounds: the Grignard reagents-formation, structure and chemical reactions. Organozinc compounds: formation and chemical reactions. Organolithium compounds: formation and chemical reactions.	

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