

GSQ/D-21

**1072**

PHYSICAL CHEMISTRY

Paper–XVI CH-302

Time : Three Hours]

[Maximum Marks : 32

**Note :** Attempt *five* questions, selecting at least *two* questions from each section Question no. 1 is compulsory.

**Compulsory Question**

1. (a) What happens to the energy of the particle in one dimensional box if the length of the box is made large? 2
- (b) How does measurement of magnetic susceptibility confirm that the formula of hypo phosphoric acid is  $H_4P_2O_6$  and not  $H_2PO_3$ ? 2
- (c) Why  $CO_2$  molecule can't give pure rotational spectra but give vibrational spectra? 2
- (d) What is the selection rule for pure rotational spectroscopy and pure rotational Raman spectroscopy? 2

**SECTION–A**

2. (a) Derive Planck's radiation law. 3
- (b) Explain the photoelectric effect, why it was not explained on the basis of Classical mechanics. 2
- (c) Write a note on Eigen values. 1

3. (a) Evaluate the commutator  
 $[x^2, d^2/dx^2]$ . 2½
- (b) Write results of a particle moving in one dimensional box. 2
- (c) To prove the Eigen values of a Hermitian Operator are all real values. 1.5
4. (a) Prove that  
 $\Delta x \cdot \Delta p_x > h/4\pi$  with the help of a wave function which represents a particle moving in one dimensional box. 3
- (b) Write a note on Optical activity, its origin and acts as constitutive property. 3
5. (a) Derive Clausius-Mosotti equation. 2
- (b) The dipole moment of HBr is 0.78D and the bond distance is 1.40 Å. Calculate the percentage ionic character of the H-Br bond. 1½
- (c) Explain the applications of magnetic susceptibility. 2½

### SECTION-B

6. (a) Why band spectrum is observed in electronic spectrum of a molecule while single line spectrum is observed in atomic spectra. 2
- (b) Calculate vibrational degree of freedom in following molecules (i) CO<sub>2</sub> (ii) C<sub>2</sub>H<sub>2</sub> (iii) Benzene (iv) He. 2
- (c) Define signal to noise ratio. 2

7. (a) Explain Isotopic effect on pure rotational spectra by taking an example of  $^1\text{HCl}$  &  $^2\text{HCl}$ . 3
- (b) Which are rotational active molecules among the following :
- (i) CO
  - (ii)  $\text{H}_2$
  - (iii)  $\text{CH}_4$
  - (iv)  $\text{H}_2\text{O}$ . 2
- (c) Which region of electromagnetic spectrum belongs to rotational spectrum? 1
8. (a) Calculate the force constant for the bond in  $\text{HCl}$  from the fact that fundamental vibration frequency is  $8.667 \times 10^{13} \text{ S}^{-1}$ . 3
- (b) Write a note on the elementary idea of vibrational frequencies of fundamental transition due to some functional groups in organic molecules. 3
9. (a) Explain O, Q & S-branches of lines in Raman spectra. 3
- (b) Write a note on the Raman effect in terms of polarizability. 3
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