Roll No	Total Pages: 3
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GSE/M-20

1481

CHEMISTRY (Inorganic Chemistry) Paper–IV (CH-104)

Time : Three Hours] [Maximum Marks : 32

Note: Attempt *five* questions in all, selecting *two* questions from each section. Question No. 1 is compulsory.

Compulsory Question

- **1.** Answer in short :
 - (i) What is Dry Ice?
 - (ii) Why Nitrogen do not form Pentahalides?
 - (iii) Why Sodium is stored in kerosine and not in HO?
 - (iv) Why inter-halogens are more reactive than Halogens?
 - (v) Draw the structure of dimer of Carboxylic acid.
 - (vi) What is p-n-p transistor?
 - (vii) What are Silicones?
 - (viii) Why bond angle in NH₃ is 107° while in NF₃ it is 102°?

 (1×8)

SECTION-A

2.	(i)	What are Intrinsic semiconductors?	(2)
	(ii)	What are the conditions for the formation of Hydro	ogen
		bond?	(2)
	(iii)	Salts of alkaline earth metals are colourless	and
		diamagnetic. Why?	(2)
3.	(1)	How does BAND theory explains metallic chara	
		of Lithium?	(2)
	(ii)	Discuss structure of BeCl ₂ in solid state and in vap	pour
		state.	(2)
	(iii)	Describe briefly dipole-dipole forces.	(2)
4.	(i)	What was the reasoning applied by N. Bartlett to c	arry
		out reaction of Xenon with PtF ₆ ?	(2)
	(ii)	Discuss structure of XeO ₂ F ₂ .	(2)
	(iii)	Why NaOH is more basic than Ba(OH) ₂ ?	(2)
5.	(i)	What are Rectifiers?	(2)
	(ii)	Why conductivity of Li ⁺¹ ion in aqeous solution is	
		less than Na ⁺ ion?	(2)
	(iii)	Explain why electron affinity of noble gases are zero.	(2)
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SECTION-B

6.	(i)	Discuss the structure of Borazine.	(2)
	(ii)	BH ₃ does not exists but dimerise to give B ₂ H ₆ . Explain.	
			(2)
	(iii)	Why BF ₃ is not hydrolysed by H ₂ O?	(2)
7.	(i)	Discuss diagonal relationship of Boron with Silicon.	(2)
	(ii)	How are cross-linked Silicones prepared?	(2)
	(iii)	Write structure of HNO ₃ and N ₂ O ₅ .	(2)
8.	(i)	Name any four oxy acids of sulphur and write	their
		structures.	(2)
	(ii)	What happens when H ₂ O ₂ reacts with	
		(a) Acidified FeSO ₄	
		(b) Acidified KMnO ₄ .	(2)
	(iii)	Why Chlorine forms ClF ₃ but fluorine does not form FCl ₃ .	
			(2)
9.	(i)	Write note on Orthosilicates and Cyclic silicates.	(2)
	(ii)	Give any three methods of preparation of B ₂ H ₆ .	(2)
	(iii)	Discuss structure of ClF ₃ .	(2)